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A NEW SYNTHESIS OF γ -ALKYLIDENE- $\Delta^{\alpha,\beta}$ -BUTENOLIDES

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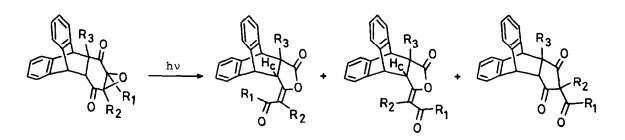
The structural unit of γ -alkylidene- $\Delta^{\alpha,\beta}$ -butenolide occurs in a variety of natural products, such as patulin,¹) protoanemonin,²) tetrenolin,³) freelingyne,⁴) and matricarialactone,⁵) possessing a wide range of biological activity. This fact has stimulated considerable research on the development of new synthetic methods for the construction of this structural unit.⁶)

We now wish to report a facile method for construction of the γ -alkylidene- $\Delta^{\alpha,\beta}$ -butenolide moiety based on the recently reported photochemical rearrangement of 2,3-epoxy-1,4-cyclohexanediones into γ -alkylidene- γ -butyrolactones,⁷⁾ combined with the retro-Diels-Alder reaction.

UV-irradiation of epoxy diketone <u>la</u>, derived from p-benzoquinone,⁸⁾ for 20 hrs in acetone, using a 100-W high pressure Hg lamp (pyrex), gave a mixture of <u>2aE</u>, <u>2aZ</u>, <u>3a</u>, and <u>la</u> in a ratio of 1:1:2.5:1 (NMR).⁹⁾ Preparative TLC separation using phosphate-buffered silica gel layer gave lactones <u>2aE</u> [10%; mp 198°(dec)] and <u>2aZ</u> [13%; mp 191°(dec)], with <u>la</u> (13%) recovered. The triketone (<u>3a</u>), very liable to decompose on TLC, was the sole product on irradiation of <u>la</u> in AcOEt¹⁰⁾ [80%; mp 163°(dec); δ_{CHO} 9.13].

Retro-Diels Alder reaction of 2aE at 230° (20 mmHg) and preparative TLC separation gave a 1:1 mixture of butenolides 4aE and 4aZ in 80% yield, along with anthracene (90%). The same ratio of the products was obtained from the reaction of 2aZ, indicating the isomerization of 4aE and 4aZ under reaction conditions.

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a: $R_1 = R_2 = R_3 = H$ b: $R_1 = R_3 = Me$, $R_2 = H$ c: $R_1 = H$, $R_2 = R_3 = Me$ $\frac{2E}{1} \qquad \frac{2Z}{1} \qquad \frac{3}{2}$

<u>4</u>E

<u>4</u>Z

This success encouraged us to apply this method to the synthesis of butenolide $\underline{4}bZ$,¹¹⁾ a key intermediate for synthesis of freelingyne, a furanosesquiterpenoid. Irradiation of <u>1b</u>, derived from 2,6-dimethyl-p-benzoquinone,¹²⁾ in acetone led to a mixture of <u>2bE</u>, <u>2bZ</u>, and <u>3b</u> in a ratio of 1:1.4:2.5. Other possible products, aldehydes <u>2c</u> and <u>3c</u>, were not detected. Preparative TLC (buffered) gave <u>2bE</u> (12%; mp 174-176°), <u>2bZ</u> (15%; mp 167-168°), and <u>3b</u>¹³⁾ (15%).

Pyrolytic treatment¹⁴⁾ of the crude photolysate gave butenolides $\underline{4}bE$ (17% from 2b; mp 88.5-89.5°, lit.¹¹⁾mp 88-89°) and $\underline{4}bZ$ (6%; mp 124-125°, lit.¹¹⁾mp 124-125°), in addition to anthracene. These two compounds have been already prepared by tedious and low-yield procedures.¹¹⁾

The stereochemistry of each of the E,Z-isomers was determined on the basis of the facts that allylic protons cis to the carbonyl function in α,β -unsaturated carbonyl compounds, or olefinic protons cis to the lactone oxygen atom resonate at a lower field than their trans counterparts.¹⁵⁾ The NMR data are shown in the Table.

		R ₁	R ₂	н _с	R ₃
<u>2</u> aE 2aZ	R ₁ =R ₂ =R ₃ =H	9.99 9.68	5.80 5.49	4.02 3.65	3.36 3.28
$\frac{2bE}{2bZ}$	$R_1 = R_3 = Me, R_2 = H$	2.28 2.15	5.95 5.45	3.54 3.03	1.20 1.23
$\frac{4}{4}aE$	R ₁ =R ₂ =R ₃ =H	9.93 10.23	8.14 7.59	6.63 6.48	6.15 5.62
<u>4</u> bе <u>4</u> bz	R ₁ =R ₃ =Me, R ₂ =H	2.31 (2.3) 2.60 (2.5)	7.96 (8.0) 7.16 (7.1)	6.13 (6.1) 5.50 (5.4)	2.06 (2.1) 2.13 (2.15)

Table. NMR chemical shifts of 2 and 4.

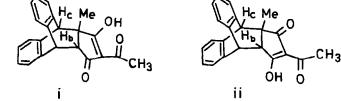
The values in parentheses donate the literature values.¹¹⁾

References and Notes

- 1) R. B. Woodward and G. Singh, J. Am. Chem. Soc., <u>71</u>, 758 (1949).
- 2) E. Shaw, J. Am. Chem. Soc., 68, 2510 (1946).
- G. G. Gallo, C. Coronelli, A. Vigevani, and G. C. Lancini, Tetrahedron, <u>25</u>, 5677 (1969).
- R. A. Massy-Westropp, G. D. Reynolds, and T. M. Spotswood, Tetrahedron Letters, 1939 (1966).
- 5) F. Bohlmann, Fortsch. Chem. Org. Naturst., 25, 5677 (1969).
- 6) Y. S. Rao, Chem. Rev., 76, 625 (1976).
- 7) T. Kitamura, T. Imagawa, and M. Kawanisi, Tetrahedron Letters, 0000 (1978).
- 8) P. Yates and P. Eaton, J. Am. Chem. Soc., 82, 4436 (1960).
- 9) Irradiation at lower temperature (about -70°) resulted in slight increase of the ratio of (2aE+2aZ)/3.
- 10) Irradiation of <u>la</u> in a variety of solvents such as benzene, CH₂Cl₂, CH₃CN,

and EtOH led to $\frac{3a}{a}$ in addition to a trace amount of 2aE and 2aZ.

- 11) C. F. Ingham, R. A. Massy-Westropp, and G. D. Reynolds, Aust. J. Chem., <u>28</u>, 2499 (1975), 27, 1477 (1974).
- 12) The epoxy diketone <u>lb</u> (mp 185-186°) was prepared by epoxydation with basic hydrogen peroxide of the AlCl₃-catalyzed Diels-Alder adduct (mp 150.5-151.5°) between 2,6-dimethy-p-benzoquinone and anthracene.
- 13) As a tautomeric mixture of i and ii (1:1); NMR: Me, 1.03(s) and 1.20(s), COCH₃, 2.20, H_b, 2.42(d) and 2.73 (d), H_c, 4.32(s) and 4.27(s), aromatic H, 6.9-7.5(m).



14) The other product <u>3</u>b remained as the still residue.

15) L. M. Jackman and S. Sternhell, Applications of Nuclear Magnetic Resonance Spectroscopy in Organic Chemistry, 2nd Ed., Pergamon Press, New York, London, p. 222 (1969).

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